



EUROPEAN PATENT APPLICATION

(43) Date of publication:

29.12.1997 Bulletin 1997/52

(51) Int Cl.⁶: H01M 6/12

(21) Application number: 97304309.4

(22) Date of filing: 19.06.1997

(84) Designated Contracting States:

AT BE CH DE DK ES FI FR GB GR IE IT LI LU MC
NL PT SE

(30) Priority: 19.06.1996 JP 157967/96

08.08.1996 JP 210160/96

12.08.1996 JP 212585/96

12.08.1996 JP 212586/96

(71) Applicant: TOSHIBA BATTERY CO., LTD.
Tokyo (JP)

(72) Inventors:

- Kaido, Hideki
Yokohama-shi, Kanagawa-ken (JP)

- Matsumoto, Kouichi
Yokohama-shi, Kanagawa-ken (JP)
- Hashimoto, Minoru
Yokohama-shi, Kanagawa-ken (JP)
- Furulchi, Mitsuru
Chigasaki-shi, Kanagawa-ken (JP)

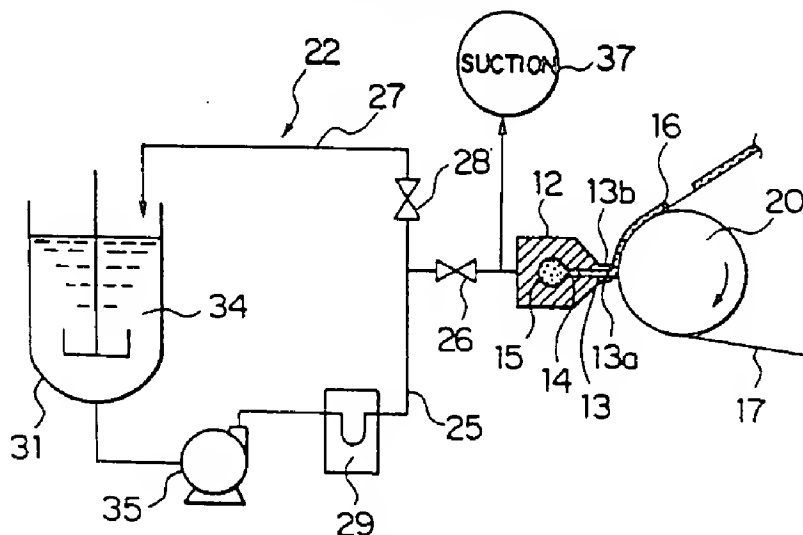
(74) Representative: Waldren, Robin Michael
MARKS & CLERK,
57-60 Lincoln's Inn Fields
London WC2A 3LS (GB)

(54) **Electrode plate for nonaqueous electrolyte battery, method of manufacture and apparatus**

(57) A method for manufacturing an electrode plate of a nonaqueous electrolyte battery comprises the steps of: running a sheet conductive base material in a first direction; and injecting an electrode material composition

containing an electrolyte from a die nozzle onto a first surface of the running sheet conductive base material to form uncoated areas at predetermined intervals on the first surface along the first direction.

FIG. 3



Description

The invention relates to a nonaqueous electrolyte battery, an electrode plate for the nonaqueous electrolyte battery, a method for manufacturing the electrode plate for the nonaqueous electrolyte battery, and an apparatus for manufacturing the electrode plate for the nonaqueous electrolyte battery.

A nonaqueous electrolyte battery having lithium as an anode active material is attracting attention as a high-energy density battery, and particularly a primary battery having manganese dioxide, carbon fluoride or thionyl chloride as a cathode active material is extensively used as power supply for a calculator or a timepiece and as a backup battery for a memory.

Besides, with miniaturization and weight reduction of a camcorder, a laptop computer, a portable phone and other various electronic equipment in these years, demands for a secondary battery with a high-energy density as a power supply for such equipment are increasing, and a lithium secondary battery having a carbon material as anode active material is being studied vigorously.

Batteries having a nonaqueous electrolyte with a main component of an organic electrolytic solution (hereinafter referred to as the nonaqueous electrolyte battery) which are represented by a lithium battery need to have a thin electrode plate because electric conductivity of the nonaqueous electrolyte is lower than an aqueous electrolyte. And, since a large reaction area is needed to obtain a large current, the cathode and anode plates are made into a sheet, and these electrode plates are wound into a roll through the intermediary of a separator to form a spiral structure. The electrode plate for such a structure is a coated type electrode which is obtained by coating a conductive base material with a coating liquid which contains an active material and a conductive material which are called as the mix.

Fig. 13 shows one example of such a battery structure.

The battery shown in Fig. 13 has a cathode plate 1 and an anode plate 2 laminated with a separator 3 which is an insulative film having a high ion transmittance therebetween, wound into a roll and housed in a container 4. And, a positive tab plate 5 is connected to a current collector at the inner periphery end of the cathode plate 1, and the positive tab plate 5 is further connected to a positive terminal 6 through a cap 6a. A negative tab plate 7 is connected to a current collector at the outer periphery end of the anode plate 2, and the negative tab plate 7 is also connected to a negative terminal 8 at the bottom of the container 4.

The cathode plate 1 and the anode plate 2 are produced by cutting to a desired length a sheet electrode plate which has both surfaces of a conductive base material coated with an electrode mix (electrode material coating agent) which mainly consists of a cathode or anode active material, a conductive agent and a binding agent. As shown in Fig. 14A to 14C, to connect the positive tab plate 5 and the negative tab plate 7 to the cathode plate 1 and the anode plate 2 respectively, it was conventional that a cross-shaped incision was formed in an electrode plate 9 (cathode plate or anode plate) and a tab plate 10 (cathode tab plate or anode tab plate), and the tab plate 10 was folded backward and crimped. But, a battery which had the tab plate 10 connected by this method had an unstable impedance. And, there were also disadvantages that the crimped part became thick and the tips of crimped part 11 ruptured the separator inducing a short circuit.

And, the tab plate is also connected to the electrode plate by welding, but the welded part on the electrode plate is required to be the base of the conductive base material without coated with the electrode mix.

To expose the base of the conductive base material, it was conventional to remove the electrode mix layer which was formed by coating and drying an electrode material coating liquid. For example, Japanese Patent Publication No. Sho 60-48865 discloses a method of removing the electrode mix layer by forming lines with a pair of knife-edges and scraping the electrode mix layers between the lines by a screw slotting cutter. And, Japanese Patent Laid-Open Publication No. Hei 2-98040 proposes a method of removing the electrode mix by contacting blades from top and bottom to the base material both surface of which are coated with the electrode mix layer. But, these methods needed an extra work of removing the electrode mix and wasted the removed electrode mix.

On the other hand, there is a method of coating the electrode mix with the base of the conductive base material left uncoated. Specifically, there is proposed a method of manufacturing a sheet electrode plate wherein a conductive agent is mixed with an electrode active material, a binding agent or the like is added thereto to prepare a paste electrode material (electrode mix) coating liquid, and this coating liquid is alternately applied and not applied to the surface of a conductive base material.

And, to produce the sheet electrode plate for electrodes as described above, there have been proposed a method of force charging a conductive mix which has a conductive agent and a binding agent kneaded with an electrode active material into a supporting material (conductive base material) while rolling, a method of extruding the kneaded electrode mix to form on both sides of a supporting material (Japanese Patent Laid-Open Publication No. Hei 4-282558), a pickup method (Japanese Patent Laid-Open Publication No. Sho 62-256365, Japanese Patent Laid-Open Publication No. Sho 63-114058), a pull-down method (Japanese Patent Laid-Open Publication No. Hei 1-267953, Japanese Patent Laid-Open Publication No. Hei 1-194265), a reverse-roll method, a gravure-roll method, a doctor blade method, and a method which uses an extrusion type injector having a slot nozzle (Japanese Patent Laid-Open Publication No. Hei

the first direction. The uncoated area is formed to weld the tab thereto to take electric power from the electrodes. The uncoated area is quite free from the electrode material composition and exposes the metal surface, so that the tab can be welded efficiently in a short time.

5 The method for manufacturing of claim 10 injects the electrode material composition onto the surface of the running sheet conductive base material to continuously decrease or increase at the predetermined ratio the coated amount per unit area along the first running direction of the sheet conductive base material.

Therefore, in manufacturing the battery, when the sheet conductive base material on which the electrode material composition is coated is wound around the shaft which is parallel to the second side of the sheet conductive base material, the distance between the two opposed electrodes, namely between the cathode and the anode, is different
10 at the center and the outer periphery of the finished battery and varies continuously from the center to the outer periphery of the battery. As a result, even if a radius of curvature is different between the center and the outer periphery of the battery, the distribution of the electrode material composition held between the cathode and the anode can be adjusted at the center and the outer periphery of the battery because the distance between the cathode and the anode is appropriately disposed, so that the charging and discharging characteristic can be made uniform at the center and the
15 outer periphery of the battery. Thus, the charging and discharging characteristic of the battery can be prevented from being degraded.

The method for manufacturing of claim 12 injects the electrode material composition from the die nozzle onto the first and second surfaces of the running sheet conductive base material in a different amount of the electrode material composition per unit area between the first surface and the second surface.

20 In manufacturing the battery, when the sheet conductive base material on which the electrode material composition is coated is wound around the shaft which is parallel to the second side of the sheet conductive base material, the amount of the electrode material composition held between the first surface and the second surface of the sheet conductive base material may be different due to a difference of the radius of curvature. But, since the coated amount of the electrode material composition is different on the first surface and the second surface of the sheet conductive base material, and this difference offsets the difference of the held amount of the electrode material composition based on
25 the difference of the radius of curvature. As a result, the charging and discharging characteristic can be made uniform at the center and the outer periphery of the battery. And, the charging and discharging characteristic of the battery can be prevented from being degraded.

The electrode plate of a nonaqueous electrolyte battery of claim 13 forms an uncoated area extending in a direction
30 of the second side intermittently with respect to a direction of the first side at a position where the first electrode material composition layer and the second electrode material composition layer are mutually opposed with respect to the first and second electrode material composition layers formed on the sheet conductive base material.

This uncoated area is formed to weld the tab thereto to take electric power from the electrodes. The uncoated area is quite free from the electrode material composition and exposes the metal surface, so that the tab can be welded
35 efficiently in a short time.

The electrode plate of a nonaqueous electrolyte battery of claim 17 has the electrode material composition layer formed on the surface of the sheet conductive base material to continuous increase or decrease the coated amount from one end to the other end along the first side of the sheet conductive base material.

Therefore, in manufacturing the battery, when the sheet conductive base material on which the electrode material composition is coated is wound around the shaft which is parallel to the second side of the sheet conductive base material, the space between the two opposed electrodes, namely between the cathode and the anode, is different at
40 the center and the outer periphery of the finished battery and varies continuously from the center to the outer periphery of the battery. As a result, even if a radius of curvature is different between the center and the outer periphery of the battery, the distribution of the electrode material composition held between the cathode and the anode can be adjusted at the center and the outer periphery of the battery because the distance between the cathode and the anode is appropriately disposed, so that the charging and discharging characteristic can be made uniform at the center and the
45 outer periphery of the battery. Thus, the charging and discharging characteristic of the battery can be prevented from being degraded.

The electrode plate of a nonaqueous electrolyte battery of claim 19 has the electrode material composition layer
50 formed on the sheet conductive base material to continuously increase or decrease the active material density in the electrode material composition layer from one end to the other end along the first side of the sheet conductive base material.

In manufacturing the battery, when the sheet conductive base material on which the electrode material composition is coated is wound around the shaft which is parallel to the second side of the sheet conductive base material, the amount of the electrode material composition held between the two opposed electrodes, namely between the cathode
55 and the anode, may be different at the center and the outer periphery of the battery due to a difference of the radius of curvature.

But, the invention has the active material density in the electrode material composition continuously varied, and

be prevented from being degraded.

Fig. 1 is a sectional view schematically showing one embodiment of the die nozzle used according to a first aspect of the invention;

Fig. 2 is a plane view showing a coated pattern of an electrode mix coating liquid formed on a conductive base material according to the first aspect of the invention;

Fig. 3 is a diagram schematically showing one embodiment of a coating device used in the method of manufacturing a sheet electrode plate according to the invention;

Fig. 4 is a diagram showing another embodiment of disposing the die nozzle of the above coating device;

Fig. 5 is a diagram showing another embodiment of an intermittent supply system for the coating liquid in the coating device used according to the invention;

Fig. 6 is a sectional view along a coated direction on a coating sheet obtained according to the invention;

Fig. 7 is a diagram schematically showing one embodiment of a coating device used for a manufacturing method of the sheet electrode plate according to the invention;

Fig. 8 is perspective view showing one embodiment of the roller press used for a manufacturing method of the sheet electrode plate according to the invention;

Fig. 9 is a diagram schematically showing one embodiment of a coating device used for a manufacturing method of the sheet electrode plate according to the invention;

Fig. 10A is a plane view showing a coating pattern of an electrode mix coating liquid coated onto a conductive base material by the coating device used for a manufacturing method of the sheet electrode plate according to the invention, and Fig. 10B is a sectional view showing the same pattern;

Fig. 11 is a sectional view showing the sheet electrode plate after pressurizing;

Fig. 12 is a diagram schematically showing one embodiment of a coating device used for a manufacturing method of the sheet electrode plate according to the invention;

Fig. 13 is a sectional view showing one embodiment of the structure of a cylindrical nonaqueous electrolyte battery; and

Fig. 14A is a diagram showing the front surface of an electrode plate to which a tab plate was connected by a conventional method, Fig. 14B is a diagram showing the rear surface of the same part, and Fig. 14C is a perspective sectional view taken on line A-A of Fig. 14A.

Fig. 1 shows one embodiment of a die nozzle used in the invention.

It is seen that a die nozzle 12 has two lips 13 mutually opposed with an appropriate space therebetween to form a land 14 and a manifold 15 for accumulating a liquid which is communicated with the land 14. An electrode material (electrode mix) composition 16 is supplied in a predetermined amount to the manifold 15 by an externally disposed coating liquid supply system (not shown), and injected from the land outlet formed at the leading end of the lips 13 through the land 14. The die nozzle 12 is disposed to have a predetermined distance from a conductive base material 17 where the tips of the lips 13 travel, and the electrode material composition 16 which is injected from the tips of the lips 13 is uniformly coated to the conductive base material 17. The material forming the die nozzle 12 is selected from materials excelling in corrosion resistance when the composition is corrosive but generally selected from a metal, an alloy, ceramics or plastics. And, a preferable space between the two lips 13, namely a preferable width of the outlet of the land 14, is 50-1200 μm , more preferably 100-800 μm , though variable depending on a viscosity (apparent viscosity) of the electrode material composition.

In applying the electrode material composition 16 by the die nozzle 12 according to the invention, the die nozzle 12 or the conductive base material 17 is traveled in a direction substantially perpendicular to the coated surface of the conductive base material 17 or to the end faces of the lips 13 of the die nozzle 12 to separate mutually; the die nozzle 12 is swung in a direction parallel to the travelling direction of the conductive base material 17; or the coating liquid is intermittently supplied to the die nozzle 12 by the composition supply system. Thus, as indicated by the coated patterns in Fig. 2, an uncoated portion 19 can be formed between electrode mix coated portions 18 at predetermined intervals in the longitudinal direction (coating direction) of the conductive base material 17.

The uncoated portion 19 may be formed in the same position or slightly deviated position on both surfaces of the conductive base material 17. And, an uncoated lug 19a is also formed on both ends in the breadth direction of the conductive base material 17 but not used for the electrode plate because the uncoated lug 19a is formed to prevent the composition from extending off the ends of the conductive base material 17 to adhere to a backup roll and cut off.

In the coated pattern described above, a length of the coated portion 18 and a length of the uncoated portion 19 can be varied as desired to conform with a size of the electrode to be cut off from the base material 17. And, the coated sheet having the coated patterns is cut in a direction (cutting lines are indicated by a broken line (a)) parallel to the longitudinal direction to conform with the size (height) of a battery and cut in the breadth direction to obtain a single electrode. At this time, the uncoated portion 19 may be cut off at its middle as indicated by a broken line (b), or may be

cut at its boundary with the coated portion 18 as indicated by a broken line (c). Besides, a coated film having the coated pattern as described above has preferably a thickness of 5-1800 μm (particularly, 50-500 μm) after drying.

The uncoated portion is preferably formed on both ends of the sheet conductive base material for a single non-aqueous electrolyte battery, namely on both ends in the longitudinal direction of the sheet conductive base material when the sheet conductive base material is cut to a length of the single nonaqueous electrolyte battery. When the uncoated portion is not formed and if a short circuit takes place between a cathode and an anode because of some trouble, e.g., when an electrode plate is deformed by a strong force which is applied from the outer periphery of the nonaqueous electrolyte battery to the internal core or when the electrode is pierced by a sharp material such as a nail, a large current flows between the cathode and the anode to cause a runaway reaction to make the battery a high temperature, possibly resulting in exploding. But, when the uncoated portion is formed on the both ends for the cathode and the anode, the runaway reaction does not occur because the cathode and the anode are directly contacted via the uncoated portion even if an external force is applied to cause deformation. Thus, the battery's temperature is slightly raised, but a possibility of explosion can be prevented.

The electrode material composition to be coated in the invention can contain an electrode active material having a particle diameter of 0.01-100 μm , a conductive agent, a binder, a solvent and the like. The electrode active material can be any compound into or from which H^+ , Li^+ , Na^+ , or K^+ can be inserted or discharged. Among others, a transition metallic oxide, transition metallic chalcogenide, carbonaceous material or the like can be used, and particularly, a lithium-containing transition metallic oxide or carbonaceous material is used preferably. It is preferable that the transition metal is mainly formed of Co, Mn, Ni, V or Fe, namely LiCoO_2 , LiNiO_2 , LiMn_2O_4 , LiCoVO_4 , LiNiVO_4 , $\text{LiCo}_{0.9}\text{Sn}_{0.1}\text{O}_2$, Fe_3O_4 , or V_2O_5 . And, the carbon material is desired to have a 002 plane spacing of 0.335-0.38 nm. and a density of 1.1-2.3 g/cm^3 , specifically graphite, petroleum coke, cresol resin calcined carbon, furan resin calcined carbon, polyacrylonitrile fiber calcined carbon, gas-phase grown carbon, or mesophase pitch calcined carbon.

The conductive agent can be any electronic conductive material which does not cause any chemical change in the formed battery. Generally, conductive materials such as natural graphite (flake graphite, scaly graphite, etc.), manufactured graphite carbon black, acetylene black, Ketjen black, carbon fiber, metal powder, or metal fiber can be used solely or as a mix of at least two of them, and it is particularly preferable to use graphite and acetylene black together.

The binder can be one member or a mix of at least two member of polysaccharide, thermoplastic resin or polymer having rubber elasticity which is hardly dissolved or swelled in an organic electrolytic solution to be used for the non-aqueous electrolyte battery. Specifically, starch, carboxymethylcellulose, hydroxypropyl cellulose, polyvinyl alcohol, polyvinyl chloride, polytetrafluoroethylene, polyvinylidene fluoride, fluororubber, ethylene-propylene-diene terpolymer (EPDM), styrene-butadiene rubber, polybutadiene, polyethylene oxide or the like can be used for the binder. These binders may be dissolved into a solvent or may be dispersed or suspended in a state of emulsion.

Besides, the solvent for kneading the electrode active material, conductive agent and binder can be water or an organic solvent or a mix of at least two organic solvents. The organic solvent is not limited to a particular one, but it is preferable to use N-methylpyrrolidone, xylene, toluene, acetone, methyl ethyl ketone, methylisobutyl ketone, cyclohexanone, ethanol, methanol, ethyl acetate, butyl acetate, methylene chloride, ethylene chloride, ethylcellosolve or the like.

In this invention, the electrode material composition is not limited to a particular composition, and generally consists of 1-50 parts by weight, preferably 1-10 parts by weight of a conductive agent and 0.1-50 parts by weight, preferably 0.1-20 parts by weight of a binder to 100 parts by weight of the electrode active material; it is preferable to add a solvent to adjust a solid content ratio to 10-80% by weight. The solvent is contained in an amount of 30-600 parts by weight. And, the electrode material composition is a liquid having an apparent viscosity of 500-100000 $\text{mPa}\cdot\text{s}$, more preferably 1000-50000 $\text{mPa}\cdot\text{s}$, at a shear rate of 13 sec^{-1} . When the composition has an apparent viscosity of less than 500 $\text{mPa}\cdot\text{s}$, it drips easily from the lip end of the die nozzle when the supply of the composition is stopped, or the coated composition has its portion facing the uncoated area deformed to flow to reach the uncoated area, and a contact angle to the surface of the base material falls very short of 90 degrees.

When the apparent viscosity exceeds 100000 $\text{mPa}\cdot\text{s}$, the die nozzle has an excessively high injection pressure, the land tends to have an instable bore (lip clearance), and a uniform coated thickness cannot be obtained accurately.

On the other hand, when the apparent viscosity is 500-100000 $\text{mPa}\cdot\text{s}$, the electrode material composition does not have any problem of deformation or leak and its contact angle can be kept at about 90 degrees. And, a uniform coated thickness can be obtained accurately.

Furthermore, the composition can be controlled to an appropriate temperature as required, and preferably to a range of 10-60°C (particularly 15-45°C), and more preferably to a range of 15-30°C (particularly 20-25°C), when the composition is coated. Tolerance is more preferably in an adjusting range of $\pm 2^\circ\text{C}$. And, it is desired that the composition and the die nozzle are adjusted to have the same temperature when the composition is coated.

The conductive base material used in this invention is not particularly limited, and may use a metallic foil of aluminum, copper, nickel or stainless steel, or a conductive film of inorganic oxide, organic macromolecular material or carbon. And, the conductive base material can be formed into various forms such as a continuous sheet, perforated sheet, or net sheet, and the continuous sheet is particularly preferable. Besides, the conductive base material has

preferably a thickness of 1-30 μm , and more preferably 5-30 μm .

In this invention, the electrode material composition is coated to front and back faces of the conductive base material sequentially or simultaneously, the conductive base material is transferred into a dry chamber to remove the solvent from the coated film, and it is pressed by a method of passing it through press rolls or the like. The drying method can be hot-air drying, infrared drying, contact drum drying or a combination thereof. To dry by the hot-air drying, a drying temperature is determined according to the composition, preferably set to 50-180°C (particularly, 50-160°C), and more preferably to 90-150°C (particularly, 90-130°C).

The coated sheet thus dried is then compressed by being passed through a pair of opposed rolls (press rolls) of a roller press, but the present invention can also pressurize the coated sheet in a state that an appropriate gap is disposed between the main faces of the rollers. At this time, assuming that the conductive base material has a thickness d and the compressed electrode plate has a target thickness D_0 , the gap size D is desired to satisfy $d \times 0.6 \leq D \leq D_0$. When the gap size D is less than 0.6 time of the thickness of the conductive base material, a very strong force is applied to the coated sheet having the uncoated areas while it is passing through the rollers, partly expanding or cutting the conductive base material. And, if the gap size D is larger than the target thickness D_0 , satisfactory pressurizing and compressing effects cannot be obtained.

The press rollers are made of a metal or rigid plastics which preferably has hardness of 80 or higher measured by a Shore D durometer. Metal press rollers may be used in pair, rigid plastics press rollers may be used in pair, or a metal press roller may be used in combination with a rigid plastics press roller. These rollers are determined to have an appropriate diameter depending on the material and thickness of the conductive base material. The rollers may be disposed at a single stage but a plurality of pairs of rollers may be disposed at multiple stages. And, the multi-stage rollers generally have a roll nip disposed in series. But, the multiple pairs of rollers may be disposed in a row, and the coated sheet may be passed in a staggered pattern through these rollers. Besides, the multi-stage rollers may have the same diameter or a different diameter.

A pressure for compressing is preferably 100-700 kg/cm in linear pressure and more preferably 200-550 kg/cm. And, the rollers are not limited to a particular temperature, but heated to temperatures ranging from room temperature to 200°C.

In applying the electrode material composition by means of the die nozzle according to the invention, the uncoated area is formed by either of the following two coating liquid supply systems. Specifically, one of the coating liquid supply systems is a gas force feeding system in which an inert gas such as air, nitrogen or argon is charged under pressure into an enclosed tank which contains the electrode material composition to supply the composition to the manifold of the die nozzle. And, in this supply system, by opening or closing a solenoid valve (injection valve) which is disposed on a supply passage, the composition is supplied to the die nozzle intermittently to form the uncoated areas at predetermined intervals in the longitudinal direction of the conductive base material.

The other coating liquid supply system in the invention is a direct supply system which supplies the composition kept in an ordinary storage tank to the die nozzle by a supply pump. And, this supply system can supply the composition to the die nozzle by intermittently switching the opening or closing direction of a cross valve which is disposed on a supply passage. The composition which passes through the cross valve is supplied to the die nozzle or returned to the storage tank according to the opened or closed direction in this supply system, but the composition is continuously discharged from the tank without being stopped.

These two supply systems intermittently supply the composition to the die nozzle and move the die nozzle as described below to form the uncoated areas, so that the bulges can be prevented from being formed at the start and end of coating. Specifically, the die nozzle is traveled in a direction substantially perpendicular to the coating surface of the conductive base material or the die nozzle is swung about an appropriate point in a direction parallel to a direction that the conductive base material is traveled to separate the tip end of the die nozzle from the coating surface of the conductive base material.

In addition, the uncoated areas can also be formed by sucking a predetermined amount from the composition in the land and manifold by a vacuum pump or the like for a predetermined duration to remove with force.

And, this intermittent coating system can continuously vary (decrease or increase) the coated amount per unit area on each coated area in the longitudinal direction. For example, in the former gas force feeding system of the coating liquid supply system, the pressure of the inert gas to be supplied into the enclosed tank is continuously varied (decreased or increased), so that the amount of the composition supplied to the die nozzle can be varied continuously. And, in the latter direct supply system of the coating liquid supply system, a servo motor or the like is disposed for the supply pump which delivers the composition to continuously vary (decrease or increase) the motor speed, so that the amount of the composition supplied to the die nozzle can be varied continuously. Thus, by decreasing or increasing the amount of the composition supplied to the die nozzle, the coated layer formed on the conductive base material becomes thin or thick, and the coating amount per unit area can be decreased or increased continuously.

According to the invention, after applying the composition with the coating amount continuously varied in the longitudinal direction of the conductive base material, the coated layer is pressurized and compressed to form an electrode

mix layer which has a uniform thickness in the longitudinal direction and a density of the electrode active material contained in the layer varied (decreased or increased) continuously.

It is preferred that a difference of the coated amount per unit area at both ends in the longitudinal direction of the coated area and a difference of the active material density in the electrode mix layer after pressuring process are each preferably 2-20% with respect to the coated amount at the end with a small amount and the active material density on the low density side. When the difference of the coated amount at both ends and the difference of the active material density are less than 2%, the continuous change of the coated amount and the active material density is substantially not effective, and when it exceeds 20%, impregnation of a portion, where the coated amount per unit area is large and the active material density after the pressuring process is high, with the electrolytic solution is poor, the battery capacity is lowered and heavily deviated, and the cycle life is made short.

In the continuous change of the coated amount, it is necessary to apply intermittently as described above to form the uncoated areas at predetermined intervals in the longitudinal direction of the conductive base material in view of work efficiency. Specifically, to decrease or increase the coated amount per unit area continuously in the longitudinal direction at a single electrode coated area, it is advantageous in view of work efficiency that the uncoated areas are disposed at predetermined intervals in the longitudinal direction of the conductive base material and the coated amount or the coated thickness is continuously varied in the longitudinal direction.

And, the invention can keep the electrode mix (electrode active material) in an optimum balance for the cathode and the anode for the nonaqueous electrolyte battery which is formed by winding a sheet electrode into a cylinder by varying the coated amount of the electrode mix per unit area on the front and back surfaces of the conductive base material to change the active material density in the electrode mix layer on the both surfaces of the electrode plate after the pressuring process. At this time, the difference of the coated amount of the electrode mix between the front and back surfaces of the conductive base material is preferably at a ratio of 2-10% with respect to the smaller coated amount. If the difference between the coated amounts is less than 2%, there is substantially no effect of changing the coated amount on both surfaces, and if it exceeds 10%, the amount of the electrode mix (electrode active material) is not well-balanced between the cathode and the anode, some portions are poorly impregnated with the electrolytic solution to lower the battery capacity, and the cycle life is shortened. For example, when LiCoO_2 is used as the cathode active material for a lithium secondary battery and the amount of the electrode active material for the cathode is larger than for the anode, all the Li ions generated from the cathode are not fully intercalated by the anode. As a result, Li metal is deposited on the surface of the anode, causing a problem in view of safety and also deteriorating the cycling property.

The cathode and the anode produced from the sheet electrode plate which is produced by the method according to the invention can be used to produce a secondary battery in the shape of a cylinder or square. A separator for separating the cathode sheet and the anode sheet can be a polyethylene film, a micro-cellular polypropylene film or a glass fiber film. And, the electrolyte is a solution consisting of as the organic solvent a solvent which is prepared by mixing at least one of aprotic organic solvents such as propylene carbonate, ethylene carbonate, butylene carbonate, dimethyl carbonate, diethyl carbonate, γ -butyrolactone, 1,2-dimethoxyethane and tetrahydroxyfuran and lithium salt soluble into the solvent, such as at least one salt, e.g., LiClO_4 , LiBF_4 , LiPF_6 , LiCF_3SO_3 , LiCF_3CO_2 or LiAsF_6 .

The invention will be described with reference to the accompanying drawings.

Fig. 3 is a diagram schematically showing the coating device used in the method of manufacturing a sheet electrode plate of the invention.

In this coating device, a die nozzle 12 is disposed such that a conductive base material 17 is continuously traveled in close contact with the surface of a revolving backup roll 20, and the tip of a lip 13 is opposed to the conductive base material 17 with a certain distance between them. The die nozzle 12 is preferably disposed so that the lip 13 is exactly square to the coating surface of the conductive base material 17, but may be disposed at another angle other than the right angle. The backup roll 20 serves to keep the space between the conductive base material 17 and the die nozzle 12 (the tip of the lip 13) and to keep the conductive base material 17 at a constant travelling speed. And, as shown in Fig. 4, the die nozzle 12 can also be disposed so that the conductive base material 17 is travelled in close contact with the neighboring roll surfaces of the backup roll 20 and a guide roll 21, and the tip of the lip 13 is kept with a predetermined space away from the conductive base material 17 at the middle of these rolls.

The die nozzle 12 has two lips 13 (an inlet lip 13a, an outlet lip 13b) which are mutually opposed with a predetermined gap therebetween, and a land 14 is formed of these lips 13. And, the die nozzle 12 has a manifold 15 which is communicated with the land 14 and keeps a liquid in it. The manifold 15 serves to relieve a change of the supplied amount of the coating liquid. An electrode material coating liquid 16 is quantitatively supplied to the manifold 15 by a coating liquid supply system 22 which is disposed outside of the die nozzle 12, and injected from the tips of the lips 13 through the land 14.

The coating liquid supply system 22 has a coating liquid tank 31 which keeps the prepared electrode material coating liquid 16, a constant rate pump 35 which continuously supplies the coating liquid 34 kept in the tank 31 to the manifold 15 of the die nozzle 12, a supply passage (supply path) 25 which connects the coating liquid tank 31 and the

was 2 m/min. And, the coating liquid was supplied to the die nozzle by employing the intermittent supply system at the same time.

Embodiment 8:

The same cathode mix coating liquid (a solid content concentration of 65% by weight and an apparent viscosity of 25000 mPa·S) as in Embodiment 2 was coated to both surfaces of an aluminum foil with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 7.

Embodiment 9:

The same cathode mix coating liquid (a solid content concentration of 70% by weight and an apparent viscosity of 40000 mPa·S) as in Embodiment 3 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 7.

Embodiment 10:

The same cathode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 4 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method 2) above. While the coating liquid was being coated, a space between the tips of the lips and the copper foil as the conductive base material was 0.3 mm, and when coating was stopped, the die nozzle was retracted at an angle of 3° in a direction opposite to the travelling direction of the copper foil, and the coating speed was 2 m/min. And, the coating liquid was supplied to the die nozzle by employing the intermittent supply system at the same time.

Embodiment 11:

The same cathode mix coating liquid (a solid content concentration of 64% by weight and an apparent viscosity of 10000 mPa·S) as in Embodiment 5 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 10.

Embodiment 12:

The same cathode mix coating liquid (a solid content concentration of 65% by weight and an apparent viscosity of 20000 mPa·S) as in Embodiment 6 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 10.

Embodiment 13:

The same cathode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 1 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method 2) above. While the coating liquid was being coated, a space between the tips of the lips and the aluminum foil as the conductive base material was 0.4 mm, and when the coating was stopped, the aluminum foil was moved backward by a distance of 5 mm in a direction perpendicular to the tip face of the die nozzle, and the coating speed was 2 m/min. And, the coating liquid was supplied to the die nozzle by employing the intermittent supply system at the same time.

Embodiment 14:

The same cathode mix coating liquid (a solid content concentration of 65% by weight and an apparent viscosity of 25000 mPa·S) as in Embodiment 2 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 13.

Embodiment 15:

The same cathode mix coating liquid (a solid content concentration of 70% by weight and an apparent viscosity of 40000 mPa·S) as in Embodiment 3 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 13.

Embodiment 16:

The same anode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 4 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method 3) above. While the coating liquid was being coated, a space between the tips of the lips and the copper foil as the conductive base material was 0.3 mm, and when the coating was stopped, the copper foil was moved backward by a distance of 5 mm in a direction perpendicular to the tip face of the die nozzle, and the coating speed was 2 m/min. And, the coating liquid was supplied to the die nozzle by employing the intermittent supply system at the same time.

Embodiment 17:

The same anode mix coating liquid (a solid content concentration of 64% by weight and an apparent viscosity of 10000 mPa·S) as in Embodiment 5 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 16.

Embodiment 18:

The same anode mix coating liquid (a solid content concentration of 68% by weight and an apparent viscosity of 20000 mPa·S) as in Embodiment 6 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 16.

Embodiment 19:

The same cathode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 1 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method of intermittently supplying the coating liquid to the die nozzle indicated in 4) above. While the coating liquid was being coated or stopped from being coated, a space between the tips of the lips and the aluminum foil as the conductive base material was 0.4 mm, and the coating speed was 2 m/min.

Embodiment 20:

The same cathode mix coating liquid (a solid content concentration of 65% by weight and an apparent viscosity of 25000 mPa·S) as in Embodiment 2 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 19.

Embodiment 21:

The same cathode mix coating liquid (a solid content concentration of 70% by weight and an apparent viscosity of 40000 mPa·S) as in Embodiment 3 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 19.

Embodiment 22:

The same anode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 4 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3

according to the method of intermittently supplying the coating liquid to the die nozzle indicated in 4) above. While the coating liquid was being coated or suspended from being coated, a space between the tips of the lips and the copper foil as the conductive base material was 0.3 mm, and the coating speed was 2 m/min.

5 Embodiment 23:

The same anode mix coating liquid (a solid content concentration of 64% by weight and an apparent viscosity of 10000 mPa·S) as in Embodiment 5 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 22.

10

Embodiment 24:

The same anode mix coating liquid (a solid content concentration of 68% by weight and an apparent viscosity of 20000 mPa·S) as in Embodiment 6 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Embodiment 22.

15

Comparative Embodiment 1:

A cathode mix coating liquid having a solid content concentration of 50% by weight and an apparent viscosity of 450 mPa·S (a shear rate of 13 sec^{-1}) was prepared by kneading LiCoO_2 , acetylene black, a fluororubber-based binder and ethyl acetate in the same way as in Embodiment 1. This coating liquid was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method indicated in 2) above in the same way as in Embodiment 7.

20

25 Comparative Embodiment 2:

A cathode mix coating liquid having a solid content concentration of 75% by weight and an apparent viscosity of 120000 mPa·S (a shear rate of 13 sec^{-1}) was prepared by kneading LiCoO_2 , acetylene black, a fluororubber-based binder and ethyl acetate in the same way as in Embodiment 1. This coating liquid was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method indicated in 2) above in the same way as in Embodiment 7.

30

Comparative Embodiment 3:

An anode mix coating liquid having a solid content concentration of 45% by weight and an apparent viscosity of 420 mPa·S (a shear rate of 13 sec^{-1}) was prepared by kneading fibrous carbon, styrene-butadiene copolymerized compound, carboxymethylcellulose and pure water in the same way as in Embodiment 4. This coating liquid was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method indicated in 2) above in the same way as in Embodiment 10.

35

40

Comparative Embodiment 4:

An anode mix coating liquid having a solid content concentration of 70% by weight and an apparent viscosity of 110000 mPa·S (a shear rate of 13 sec^{-1}) was prepared by kneading fibrous carbon, styrene-butadiene copolymerized compound, carboxymethylcellulose and pure water in the same way as in Embodiment 4. This coating liquid was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 3 according to the method indicated in 2) above in the same way as in Embodiment 10.

45

50

Comparative Embodiment 5:

The same cathode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 1 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by a reverse roll method which stops coating when the backup roll is separated. The coating speed was 2 m/min.

55

Comparative Embodiment 6:

5 The same cathode mix coating liquid (a solid content concentration of 65% by weight and an apparent viscosity of 25000 mPa·S) as in Embodiment 2 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

Comparative Embodiment 7:

10 The same cathode mix coating liquid (a solid content concentration of 70% by weight and an apparent viscosity of 40000 mPa·S) as in Embodiment 3 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

15 Comparative Embodiment 8:

The same cathode mix coating liquid (a solid content concentration of 50% by weight and an apparent viscosity of 450 mPa·S) as in Comparative Embodiment 1 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

Comparative Embodiment 9:

25 The same cathode mix coating liquid (a solid content concentration of 75% by weight and an apparent viscosity of 120000 mPa·S) as in Comparative Embodiment 2 was coated to both surfaces of an aluminum foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

Comparative Embodiment 10:

30 The same anode mix coating liquid (a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S) as in Embodiment 4 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

35 Comparative Embodiment 11:

The same anode mix coating liquid (a solid content concentration of 64% by weight and an apparent viscosity of 10000 mPa·S) as in Embodiment 5 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

Comparative Embodiment 12:

45 The same anode mix coating liquid (a solid content concentration of 68% by weight and an apparent viscosity of 20000 mPa·S) as in Embodiment 6 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

50 Comparative Embodiment 13:

The same anode mix coating liquid (a solid content concentration of 45% by weight and an apparent viscosity of 420 mPa·S) as in Comparative Embodiment 3 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

Comparative Embodiment 14:

The same anode mix coating liquid (a solid content concentration of 70% by weight and an apparent viscosity of 110000 mPa·S) as in Comparative Embodiment 4 was coated to both surfaces of a copper foil one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method as in Comparative Embodiment 5.

The coated sheets for cathode and the coated sheets for anode prepared in Embodiments 1 to 24 and Comparative Embodiments 1 to 14 were measured for each coated thickness to examine uniformity and stability of the coated thickness. Specifically, referring to a sectional view of the coated sheet shown in Fig. 6, when it was assumed that a coated thickness at the center of the coated area where a coated layer 32 had a stable thickness was δ_C , a difference $\delta_{C-S}(\delta_C - \delta_S)$ between the value (δ_C) and a coated thickness δ_S at a position 1 mm from the position where coating was started, a difference $\delta_{C-E}(\delta_C - \delta_E)$ between the value (δ_C) and a coated thickness δ_E at a position 1 mm short of terminating the coating, a length L_S from the start of coating until the coated thickness became uniform, and a length L_E from the point where the coated thickness was uniform to the point where the coating was terminated were measured. In the drawing, reference numeral 33 denotes an aluminum foil or a copper foil. And, a direction that the liquid is coated is indicated by an arrow. The measured results are shown in Table 1 and Table 2.

Table 1

	Embodiments																							
	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24
$\delta_{c-s}(\delta_c - \delta_s)$ (μm)	7	5	3	2	2	2	6	6	3	3	2	2	7	5	2	2	3	3	8	6	5	8	6	7
$\delta_{c-x}(\delta_c - \delta_x)$ (μm)	5	8	6	8	5	3	6	7	6	7	5	4	6	7	5	7	5	4	7	4	3	4	3	6
L_s (mm)	3	2	2	3	2	2	3	2	3	3	2	2	3	3	2	3	3	2	3	4	4	3	4	3
L_R (mm)	3	3	3	2	2	2	3	3	2	3	2	3	2	3	2	3	2	3	3	3	4	4	4	4

Table 2

	Comparative Embodiments													
	1	2	3	4	5	6	7	8	9	10	11	12	13	14
$\delta_{c-s}(\delta_c - \delta_s)$ (μm)	47	-23	34	-25	-18	-21	-25	19	-37	-22	-24	-28	21	-31
$\delta_{c-x}(\delta_c - \delta_x)$ (μm)	23	-4	28	-2	-8	-8	-10	16	-25	-16	-15	-18	19	-24
L_s (mm)	16	10	23	12	17	24	25	22	16	13	21	26	24	19
L_R (mm)	18	5	28	10	6	15	4	19	6	5	7	7	26	6

It was apparent from Embodiments 1 to 24 that each coated area had a uniform and stable coated thickness from the point immediately after starting to apply the coating liquid to the point the coating was terminated, and the obtained coated sheet had good uncoated areas completely free from the electrode mix coating liquid. And, electrodes were prepared from the prepared sheets and used to produce nonaqueous electrolyte batteries by a conventional procedure.

The batteries produced were good in performance and safety.

On the other hand, the coated sheets produced in Comparative Embodiments 1 to 14 had the uncoated areas in a good condition, but the coating liquid was not clearly stopped at the boundary between the coated area and the uncoated area, causing the coated layer to bulge at the end of the coated area or a coated thickness to be extremely thin. As a result, when the obtained coated sheet was compression-molded by a roller press, the coated sheet having the bulged coated layer had the conductive base material cut. And, when the coated sheet having an extremely thin coated layer was used to produce a battery, its performance and safety were too low to practically use for the battery.

As described above, it is apparent that the invention can efficiently form the good uncoated areas, where there is no electrode mix coating liquid, to produce a sheet electrode plate having good characteristics. And, since the uncoated areas are formed when the coating liquid is coated, it is not necessary to peel the electrode mix layer to weld a tab plate, the number of manhour is decreased to save labor, and efficiency of the coating process can be improved.

Now, the invention will be described with reference to the accompanying drawings.

Fig. 7 is a diagram schematically showing a coating device used for a manufacturing method of the sheet electrode plate according to the invention.

In this coating device, a die nozzle 12 is disposed such that a conductive base material 17 is moved continuously in close contact with the roll surface of a revolving backup roll 20, and the tips of lips 13 are kept a predetermined space away from the conductive base material 17. The die nozzle 12 has two lips 13 (an inlet lip 13a, an outlet lip 13b) which are mutually opposed with an appropriate gap therebetween, and a land 14 is formed of these lips 13. And, the die nozzle 12 has a manifold 15 which is communicated with the land 14 and keeps the coating liquid in it. The manifold 15 serves to relieve a change of the supplied amount of the coating liquid. An electrode mix coating liquid 16 is supplied to the manifold 15 by a coating liquid supply system which is disposed outside of the die nozzle 12, and injected from the outlet formed at the tips of the lips 13 through the land 14 to be coated onto the conductive base material 17.

The coating liquid supply system has a coating liquid tank 31 which keeps the electrode mix coating liquid 16, a liquid feeding pump 35 which feeds the coating liquid 16 kept in the tank 31, a supply passage (supply path) 25 which connects the coating liquid tank 31 and the die nozzle 12, a return passage 27 which is branched from the supply passage 25, and a three-way valve 36 which is disposed at a branch joint of the return passage. And, the prepared electrode mix coating liquid 16 is pumped out of the coating liquid tank 31 by the liquid feeding pump 35 and supplied to the die nozzle 12 through the supply passage 25. This supply system switches an open-close direction of the three-way valve 36 from a direction to supply to the die nozzle 12 to a direction to the return passage 27 with predetermined timing to supply intermittently the coating liquid to the die nozzle 12, and when the supply is stopped, the coating liquid 16 is returned to the coating liquid tank 31 through the return passage 27. Reference numeral 29 denotes a flow meter and 37 a suction mechanism in Fig. 7. The suction mechanism 37 is disposed to seal the tank 31 under pressure and to instantaneously release the pressure remained in the manifold 15 when the uncoated area is formed by simply opening and closing the three-way valve 36 to stop supplying the coating liquid to the die nozzle 12.

By the coating device having the above coating liquid supply system, the electrode mix coating liquid 16 is coated sequentially or simultaneously to front and back surfaces of the conductive base material 17 and dried in hot air. Then, a coated sheet is compressed under predetermined pressure by being passed through a roller press to be described below.

As shown in Fig. 8, the roller press comprises a pair of press rollers 38 disposed to oppose mutually and housings 39 which include a pressurizing mechanism and support both ends of the rotation shaft of each roller 38, and a gap spacer 40 which has a thickness equal to or larger than 0.5 time of a thickness d of the conductive base material 17 and equal to or less than a target thickness D_0 is held between the respective housings 39 which are disposed to oppose vertically. Thus, a space (gap) equivalent to the thickness of the gap spacer 40 is formed between the main surfaces of the rollers 38, and a coated sheet S which has uncoated areas 19 formed at predetermined intervals in the longitudinal direction of the conductive base material 17 is pressurized and compressed by being passed between the main surfaces of the rollers 38. Reference numeral 18 denotes a coated area and R a delivery roll.

In this way, the coated sheet S is compressed without stretching or cutting the conductive base material 17, and a sheet electrode plate having an electrode mix layer with a high activating material charging density is produced.

Now, the invention will be described in further detail referring to specific embodiments, but it is to be understood that the invention is not limited to the following embodiments unless they depart from the objects of the invention.

Embodiment 25:

A slurry-like coating liquid having a solid content concentration of 60% by weight and an apparent viscosity of 3000

mPa·S (a shear rate of 13 sec^{-1}) was prepared by mixing 90 parts by weight of LiCoO_2 as the cathode active material and 6 parts by weight of acetylene black as the conductive agent, adding 2 parts by weight of a fluororubber-based binder as the binding agent, adding ethyl acetate as the solvent, and kneading them.

5 This coating liquid was coated to both surfaces of an aluminum foil having a thickness of $15 \mu\text{m}$ one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by means of the coating device shown in Fig. 9 with the tips of the lips of the die nozzle separated in a direction perpendicular to the coating surface of the aluminum foil. After drying in hot air, the obtained coating sheet (electrode sheet with a width of 100 mm and a thickness of $260 \mu\text{m}$) was compressed by being passed between the metal-metal rollers having a diameter of 200 mm of the roller press shown in Fig. 10 to produce a cathode sheet having a thickness of about $185 \mu\text{m}$ (a target thickness of $185 \mu\text{m}$). At this time, compression was made under conditions of a linear pressure of 500 kg/cm, a speed of 2 m/min, a roller temperature of 25°C , and a gap spacer thickness (equal to the gap D between the rollers) of $15 \mu\text{m}$.

Embodiment 26:

15 A cathode sheet was produced by the same procedure as in Embodiment 25 except that the line pressure at pressing (pressurizing) was changed to 750 kg/cm as shown in Table 3.

Embodiment 27:

20 A cathode sheet was produced by the same procedure as in Embodiment 25 except that the line pressure at pressing was changed to 1000 kg/cm as shown in Table 3.

Embodiment 28:

25 A cathode sheet was produced by the same procedure as in Embodiment 25 except that the line pressure at pressing was changed to 300 kg/cm and the roller temperature was changed to 60°C as shown in Table 3.

Embodiment 29:

30 A cathode sheet was produced by the same procedure as in Embodiment 28 except that the line pressure at pressing was changed to 500 kg/cm as shown in Table 3.

Embodiment 30:

35 A cathode sheet was produced by the same procedure as in Embodiment 28 except that the line pressure at pressing was changed to 750 kg/cm as shown in Table 3.

Embodiment 31:

40 A cathode sheet was produced by the same procedure as in Embodiment 28 except that the gap spacer thickness, namely the gap D between the rollers, was changed to $12 \mu\text{m}$ as shown in Table 3.

Embodiment 32:

45 A cathode sheet was produced by the same procedure as in Embodiment 31 except that the line pressure at pressing was changed to 500 kg/cm as shown in Table 3.

Embodiment 33:

50 A cathode sheet was produced by the same procedure as in Embodiment 32 except that the gap spacer thickness, namely the gap D between the rollers, was changed to $30 \mu\text{m}$ as shown in Table 3.

Embodiment 34:

55 A cathode sheet having a thickness of about $180 \mu\text{m}$ (a target thickness of $180 \mu\text{m}$) was produced by the same procedure as in Embodiment 29 except that the aluminum foil had a thickness of $10 \mu\text{m}$ and the gap spacer thickness, namely the gap D between the rollers, was changed to $10 \mu\text{m}$ as shown in Table 3.

Embodiment 35:

A cathode sheet having a thickness of about 200 μm was produced by the same procedure as in Embodiment 29 except that the aluminum foil had a thickness of 30 μm and the gap spacer thickness, namely the gap D between the rollers, was changed to 30 μm as shown in Table 3.

Embodiment 36:

A cathode sheet having a thickness of about 200 μm was produced by the same procedure as in Embodiment 35 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 19 μm as shown in Table 3.

Embodiment 37:

An anode mix coating liquid having a solid content concentration of 60% by weight and an apparent viscosity of 3000 mPa·S (a shear rate of 13 sec^{-1}) was prepared by mixing 97 parts by weight of fibrous carbonaceous material as the anode active material, 2 parts by weight of a copolymerized compound of styrene and butadiene as the binding agent and 1 part by weight of carboxymethylcellulose as the thickening agent, adding pure water as the solvent, and kneading them.

This slurry-like coating liquid was coated to both surfaces of a copper foil having a thickness of 12 μm one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction by the same method employed to apply the above-described cathode mix coating liquid, and dried in hot air. The obtained coated sheet was compressed by being passed between the metal-metal rollers having a diameter of 200 mm in the same way as in Embodiment 1 to produce an anode sheet having a thickness of about 184 μm (a target thickness of 184 μm). At this time, compression was made under conditions of a linear pressure of 300 kg/cm, a speed of 2 m/min, a roller temperature of 25°C, and a gap spacer thickness of 12 μm .

Embodiment 38:

An anode sheet was produced by the same procedure as in Embodiment 37 except that the line pressure at pressing was changed to 500 kg/cm as shown in Table 3.

Embodiment 39:

An anode sheet was produced by the same procedure as in Embodiment 37 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 7 μm as shown in Table 3.

Embodiment 40:

An anode sheet was produced by the same procedure as in Embodiment 39 except that the line pressure at pressing was changed to 500 kg/cm as shown in Table 3.

Embodiment 41:

An anode sheet was produced by the same procedure as in Embodiment 38 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 24 μm as shown in Table 3.

Embodiment 42:

An anode sheet having a thickness of about 203 μm (a target thickness of 203 μm) was produced by the same procedure as in Embodiment 38 except that the copper foil was changed to have a thickness of 30 μm and the gap spacer thickness, namely the gap D between the rollers, was changed to 30 μm as shown in Table 3.

Embodiment 43:

An anode sheet was produced by the same procedure as in Embodiment 42 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 18 μm as shown in Table 3.

Comparative Embodiment 15:

A cathode sheet was produced by the same procedure as in Embodiment 29 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 6 μm as shown in Table 3.

Comparative Embodiment 16:

A cathode sheet was produced by the same procedure as in Embodiment 34 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 5 μm as shown in Table 3.

Comparative Embodiment 17:

A cathode sheet was produced by the same procedure as in Embodiment 35 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 12 μm as shown in Table 3.

Comparative Embodiment 18:

A cathode sheet was produced by the same procedure as in Embodiment 29 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 190 μm as shown in Table 3.

Comparative Embodiment 19:

An anode sheet was produced by the same procedure as in Embodiment 36 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 5 μm as shown in Table 3.

Comparative Embodiment 20:

An anode sheet was produced by the same procedure as in Embodiment 42 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 12 μm as shown in Table 3.

Comparative Embodiment 21:

An anode sheet was produced by the same procedure as in Embodiment 36 except that the gap spacer thickness, namely the gap D between the rollers, was changed to 190 μm as shown in Table 3.

In Embodiments 25 to 43 and Comparative Embodiments 15 to 21, the compressed electrode sheets (the number of pressing times: 1, 2 to 3, and 4 or more) were examined for a thickness and a condition of the base material. The results are shown in Table 3.

Table 3

	Embodiments								
	25	26	27	28	29	30	31	32	33
Base material and thickness (μm)	Al 15	Al 15	Al 15	Al 15	Al 15	Al 15	Al 15	Al 15	Al 15
Coated sheet thickness (μm)	260	260	260	260	260	260	260	260	260
D (μm)	15	15	15	15	15	15	12	12	30
Roller temperature ($^{\circ}\text{C}$)	25	25	25	60	60	60	60	60	60
Line pressure (kg/cm)	500	750	1000	300	500	750	300	500	500
Target thickness (μm)	185	185	185	185	185	185	185	185	185
Thickness after pressing (μm)									
1st passing	208	206	202	201	198	198	198	194	210
2nd passing								185	204
3rd passing	195	194	194	189	185	184	187		
4th passing							184		
5th passing	185	184	184	185					
8th passing									
10th or 11th passing									
State of base material	Normal	Normal	Normal	Normal	Normal	Normal	Normal	Normal	Normal

	Comparative Embodiments						
	15	16	17	18	19	20	21
Base material and thickness (μm)	Al 15	Al 10	Al 30	Al 15	Cu 12	Cu 30	Cu 12
Coated sheet thickness (μm)	260	255	286	260	220	240	220
D (μm)	6	5	12	190	5	12	190
Roller temperature ($^{\circ}\text{C}$)	60	60	60	60	25	25	25
Line pressure (kg/cm)	500	500	500	500	500	500	500
Target thickness (μm)	195	180	200	185	184	203	184
Thickness after pressing (μm)							
1st passing	191	190	210	240	183	200	211
2nd passing	183	180	198				
3rd passing				237			209
4th passing							
5th passing							
8th passing							
10th or 11th passing					234		208
State of base material	Cut-off	Cut-off	Stretch-ed	Nor-mal	Cut-off	Stretch-ed	Nor-mal

It is apparent from Table 3 that Embodiments 25 to 43 could produce a sheet electrode plate having a predetermined thickness without causing a stretch or cut in the base material in the pressing process. On the other hand, Comparative Embodiments 15 to 17, 19 and 20 could not continue the pressing process because the base material was stretched or cutoff when it was pressed one time or two times. And, the base material was not stretched or cut in Comparative Embodiments 18 and 21 but pressing of 10 times or more was insufficient to provide a pressing effect. And, an electrode

sheet having a desired thickness could not be obtained.

As apparent from the above description, the invention can compress effectively the coated sheet which is provided with the uncoated areas for connection of the tab plate in the electrode mix applying process by the roller press with a gap formed between the rollers without causing a stretch or cutoff in the conductive base material to produce a sheet electrode plate having an enhanced charging density of the active material. And, an electrode made of the obtained sheet electrode plate can be wound into the shape of a roll through the intermediary of a separator to obtain a non-aqueous electrolyte battery.

Now, the invention will be described with reference to the accompanying drawings.

Fig. 9 is a diagram schematically showing a coating device used for a manufacturing method of the sheet electrode plate according to the invention.

In this supply system, an open-close direction of the three-way valve 36 is switched from a direction to supply to the die nozzle 12 to a direction to the return passage 27 with predetermined timing to supply intermittently the coating liquid to the die nozzle 12, and when the supply is stopped, the coating liquid 16 is returned to the coating liquid tank 31 through the return passage 27. And, when the coating liquid is supplied to the die nozzle 12, it is controlled to change (decrease or increase) continuously the running speed of a servo motor and to decrease or increase continuously the delivery amount (supply amount) of the coating liquid 16 by the liquid feeding pump 35.

By the coating device having the above coating liquid supply system, the electrode material coating liquid 16 is coated sequentially or simultaneously to front and back surfaces of the conductive base material 17 to form an uncoated area 19 at predetermined intervals in the longitudinal direction as shown in Figs. 10A, 10B. And, there is obtained a coated sheet which has on each coated area 16a, 16b a coated amount per unit area, namely thickness of a coated layer 16a, 16b decreased or increased continuously along the longitudinal direction (Fig. 10 shows that the coated layer thickness is continuously decreased at a predetermined ratio along the coated direction indicated by the arrow).

Then, the coated sheet is dried in hot air and passed through a pair or multiple pairs of press rollers, thereby compressed under predetermined pressure. Thus, an electrode mix layer 16 which has a uniform thickness along the coated direction with the electrode active material density present in the layer changed (decreased or increased) continuously as shown in Fig. 11.

Now, the invention will be described in further detail referring to specific embodiments.

Embodiment 44:

A slurry-like coating liquid having a solid content concentration of 60% by weight was prepared by mixing 90 parts by weight of LiCoO_2 as the cathode active material and 5 parts by weight of acetylene black as the conductive agent, adding 5 parts by weight of a fluororubber-based binder as the binding agent, adding ethyl acetate as the solvent, and kneading them. This coating liquid was coated to both surfaces of an aluminum foil having a thickness of 20 μm one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction according to a direct supply method by a liquid feeding pump also using a three-way valve with the coated amount per unit area increased or decreased (increased on the front surface, decreased on the back surface) continuously in a range of 259-286 g/m^2 (a difference of coated amount was 10.4% with the minimum value as reference) as shown in Table 4 on both ends in the longitudinal direction of the coated area in a single pattern. Then, after drying in hot air, the obtained coated sheet was compressed by being passed between the press rollers, so that the electrode mix layer on one surface became 85 μm in thickness to produce a cathode sheet having a thickness of 190 μm which had the active material density inclined in the longitudinal direction. As shown in Table 4, the active material density in the electrode mix layer was 2.85-3.14 g/cm^3 (a difference of the active material density was 10.2% with the minimum value as reference) on both ends in the longitudinal direction of the electrode in one pattern.

A slurry-like coating liquid having a solid content concentration of 60% by weight was prepared by mixing 85 parts by weight of mesophase pitch carbon fiber as the anode active material, 5 parts by weight of acetylene black as the conductive agent and 5 parts by weight of graphite, adding 5 parts by weight of styrene butadiene rubber as the binding agent, adding water as the solvent, and kneading them. This coating liquid was coated to both surfaces of a copper foil having a thickness of 20 μm one surface at a time with uncoated areas formed at predetermined intervals in the longitudinal direction according to a gas force feeding system which fed pressurized air into the enclosed tank to pressurize. And, by continuously varying the pressure of air fed into the enclosed tank, the coating liquid was applied to the copper foil with one surface at a time while increasing or decreasing continuously the coated amount per unit area on both ends in the longitudinal direction of the coated area of one pattern in a range of 110-122 g/m^2 (a difference of 10.9%) as indicated in Table 4. Then, after drying in hot air, the obtained coated sheet was compressed by being passed between the press rollers, so that the electrode mix layer on one surface became 85 μm in thickness to produce an anode sheet having a thickness of 190 μm which had the active material density of 1.29-1.43 g/cm^3 (a difference of 10.9%) inclined in the longitudinal direction as shown in Table 4.

Then, the produced cathode and anode sheets were cut to each length (the cathode had an electrode length of

Comparative Embodiment 22:

For the cathode and anode, the coating liquid was coated in a predetermined amount by using the same electrode mix and the same means as in Embodiment 44 without varying the coating amount in one pattern. The obtained coated sheet was dried in hot air and compressed by the press rollers to produce cathode and anode sheets having a predetermined (uniform) active material density (a difference of active material density was 0%) in the longitudinal direction. Then, the cathode and anode sheets thus produced were cut to the lengths specified in Table 4 and wound into the shape of a roll through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Comparative Embodiment 23:

The cathode and anode sheets produced in Embodiment 44 were each cut to the same length as in Embodiment 44, and wound into the shape of a roll with a micro-cellular polypropylene film as the separator involved to have the end section with a larger active material density inside the wound coil to produce a cylindrical battery.

Table 4

		Embodiments												Comparative Embodiments	
		44		45		46		47		22					
Coating amount in one pattern (g/m ²)	Min.	259	110	249	107	242	105	242	105	272	116				
	Max.	286	122	298	128	302	130	302	130	272	116				
Difference of coating amount in one pattern (%)		10.4	10.9	19.7	19.6	24.8	23.8	24.8	23.8	0.0	0.0				
Active material density in one pattern (g/m ²)	Min.	2.85	1.29	2.73	1.26	2.66	1.23	2.58	1.19	3.00	1.36				
	Max.	3.14	1.43	3.27	1.50	3.32	1.53	3.20	1.47	3.00	1.36				
Difference of active material density in one pattern (%)		10.2	10.9	19.8	19.0	24.8	24.4	24.0	23.5	0.0	0.0				
Thickness of electrode mix layer on one surface (μm)		85	85	85	85	85	85	88	88	85	85				
Length of electrode housed in battery (mm)		880	900	880	900	880	900	850	870	880	900				

The cylindrical batteries produced in Embodiments 44 to 47 and Comparative Embodiments 22 and 23 were tested for an overcharge (3C-15V) and a discharge capacity and deviations. And, they were also tested for a charging and discharging cycle to measure the number of charging and discharging times (cycle life) until the capacity becomes 80% of the initial value. The results of the overcharge test were indicated by ⊙ for excellent, o for good, and x for bad. And, the discharge capacity was indicated with the battery capacity obtained in Comparative Embodiment 22 as reference. These test results are shown in Table 5.

Table 5

	Embodiments				Comparative Embodiments	
	44	45	46	47	22	23
Overcharge test	⊙	⊙	o	⊙	o	x
Comparison of capacity (%)	+12	+7	-6	-11	±0	-8
Deviation of capacity (%)	±1	±3	±7	±4	±5	±13
Cycle life (Frequency)	1200	950	500	1100	800	10

It is apparent from Table 5 that safety of the battery against overcharge was improved and its capacity was increased, and deviations in the capacity were decreased in Embodiments 44 to 47 wherein the sheet electrode having the coating amount per unit area at each electrode coating section continuously varied and the active material density in the electrode mix layer after compressing treatment also varied slantingly. And, the inclination of the coating amount and active material density at the electrode coating section in one pattern is preferably configured to decrease from the outside to the inside of a spiral structure. But, in Embodiment 46 wherein a difference between the minimum and maximum values of the coating amount and the active material density exceeds 20% of the minimum value, the electrolytic solution is poor to permeate at the outer periphery of the spiral structure having a high active material density, resulting in decreasing the capacity, increasing variations in capacity, and shortening the cycle life. Besides, when compressibility in the pressurizing treatment was lowered to prevent the active material density on the high density side from becoming excessively high so that the electrolytic solution permeates fully at the outer periphery of the spiral structure as in Embodiment 47, the electrode was made thick and the electrode to be housed in the battery become short. As a result, it is seen that the capacity is decreased substantially because the active material amount is decreased.

It is apparent from the above description that the sheet electrode plate can be produced with the uncoated areas formed at predetermined intervals in the longitudinal direction and the active material density in the electrode mix layer varied continuously in each electrode coated section between the uncoated areas by the method of the present invention. And, the electrode produced from the sheet electrode plate is wound into a roll through the intermediary of the separator and used as the cathode and/or anode, so that there can be obtained a nonaqueous electrolyte battery having excellent safety, high capacity, a less deviation of the discharge capacity, and an improved charging and discharging cycle.

The invention will be described with reference to the accompanying drawings.

Fig. 12 is a diagram schematically showing a coating device used for a manufacturing method of the sheet electrode plate according to the invention.

In this coating device, a die nozzle 12 is disposed such that a conductive base material 17 is moved continuously in close contact with the roll surface of a revolving backup roll 20, and the tips of lips 13 are kept a predetermined space away from the conductive base material 17. The die nozzle 12 has two lips 13 (an inlet lip 13a, an outlet lip 13b) which are mutually opposed with an appropriate gap therebetween, and a land 14 is formed of these lips 13. And, the die nozzle 12 has a manifold 15 which is communicated with the land 14 and keeps the coating liquid in it. An electrode mix coating liquid 16 is supplied to the manifold 15 by a coating liquid supply system (not shown) which is disposed outside of the die nozzle 12, and injected from the outlet formed at the tips of the lips 13 through the land 14 to be coated onto the conductive base material 17.

By the coating device, the electrode mix coating liquid 16 is coated with the coating amount per unit area, namely the coated thickness, varied on front and back surfaces of the conductive base material 17. And, the coated sheet is dried in hot air and passed through a pair of multiple pairs of press rollers, thereby compressed under predetermined pressure. Thus, the electrode mix layers on the front and back surfaces have a different thickness in proportion to the coated amount of the electrode mix coating liquid 16, and therefore, the electrodes formed have a different active material density in the layers.

Now, the invention will be described in further detail referring to specific embodiments.

Embodiment 48:

A slurry-like coating liquid having a solid content concentration of 60% by weight was prepared by mixing 90 parts by weight of LiCoO_2 as the cathode active material and 5 parts by weight of acetylene black as the conductive agent, adding 5 parts by weight of a fluororubber-based binder as the binding agent, adding ethyl acetate as the solvent, and kneading them. This coating liquid was coated to both surfaces of an aluminum foil having a thickness of 20 μm one surface at a time by the coating device shown in Fig. 14 and with the same coating amount (per unit area) on the front and back surfaces as shown in Table 6. Then, the obtained coated sheet was dried in hot air and compressed by the press rollers to produce a cathode sheet having a thickness of 200 μm .

A slurry-like coating liquid having a solid content concentration of 60% by weight was prepared by mixing 65 parts by weight of mesophase pitch carbon fiber as the anode active material, 5 parts by weight of acetylene black as the conductive agent and 5 parts by weight of graphite, adding 5 parts by weight of styrene butadiene rubber as the binding agent, adding water as the solvent, and kneading them. This coating liquid was coated to both surfaces of a copper foil having a thickness of 20 μm one surface at a time by the same method as for the cathode mix coating liquid with the coating amounts on the front and back surfaces changed by 4.9% (with respect to the coated amount onto the front surface) as shown in Table 6. The obtained coated sheet was dried in hot air and compressed by the press rollers to produce an anode sheet having a thickness of 200 μm .

Then, the cathode and anode sheets were cut to a predetermined size, and wound into a roll through the intermediary of a micro-cellular polypropylene film as the separator with the front surfaces of the respective foils inside and the back surfaces outside to produce a cylindrical battery.

Embodiment 49:

Cathode and anode electrode sheets were respectively produced using the same electrode mix coating liquid by the same means as in Embodiment 48 with the coating amount for the cathode changed by 4.9% between the front and back surfaces as shown in Table 6. Then, these electrode sheets were wound into a roll in the same way as in Embodiment 48 through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Embodiment 50:

Cathode and anode sheets were respectively produced using the same electrode mix coating liquid by the same means as in Embodiment 48 with the coating amount for the cathode changed by 2.6% and for the anode changed by 2.5% between the front and back surfaces as shown in Table 6. Then, these electrode sheets were wound into a roll in the same way as in Embodiment 48 through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Embodiment 51:

Cathode and anode sheets were respectively produced using the same electrode mix coating liquid by the same means as in Embodiment 48 with the coating amount for the cathode changed by 7.1% and for the anode changed by 7.2% between the front and back surfaces as shown in Table 6. Then, these electrode sheets were wound into a roll in the same way as in Embodiment 48 through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Comparative Embodiment 24:

Cathode and anode sheets were respectively produced using the same electrode mix coating liquid by the same means as in Embodiment 48 with the coating amount same for the cathode and the anode between the front and back surfaces as shown in Table 6. Then, these electrode sheets were wound into a roll in the same way as in Embodiment 48 through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Comparative Embodiment 25:

Cathode and anode sheets were respectively produced using the same electrode mix coating liquid by the same means as in Embodiment 48 with the coating amount for the anode changed by 14.0% between the front and back surfaces as shown in Table 6. Then, these electrode sheets were wound into a roll in the same way as in Embodiment 48 through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Comparative Embodiment 26:

Cathode and anode sheets were respectively produced using the same electrode mix coating liquid by the same means as in Embodiment 48 with the coating amount for the cathode changed by 14.2% between the front and back surfaces as shown in Table 6. Then, these electrode sheets were wound into a roll in the same way as in Embodiment 48 through the intermediary of a micro-cellular polypropylene film as the separator to produce a cylindrical battery.

Table 7 shows the ratio of electrode mix coating amount between the cathode and the anode which are mutually opposed with the separator therebetween in the batteries produced in Embodiments 48 to 51 and Comparative Embodiments 24 to 26. The optimum value of the ratio of coating amounts in the flat plate state is 2.50 (cathode/anode) corresponding to the ratio in Comparative Embodiment 24.

Table 6

	Embodiments				Comparative Embodiments		
	48	49	50	51	25	26	27
Cathode							
Coated amount on front surface (g/m ²)	300	293	296	289	300	300	277
Coated amount on back surface (g/m ²)	300	308	304	311	300	300	323
Difference of coated amount between front and back surfaces (%)	0.0	4.9	2.6	7.1	0.0	0.0	14.2
Anode							
Coated amount on front surface (g/m ²)	117	120	119	116	120	111	120
Coated amount on back surface (g/m ²)	123	120	122	125	120	129	120
Difference of coated amount between front and back surfaces (%)	4.9	0.0	2.5	7.2	0.0	14.0	0.0

Table 7

	Embodiments				Comparative Embodiments		
	48	49	50	51	25	26	27
Opposed electrodes							
Coated amount for cathode's inside (front surface) (g/m ²)	300	293	296	289	300	300	277
Coated amount for anode's outside (back surface) (g/m ²)	123	120	122	125	120	129	120
Ratio of coated amounts (cathode/anode)	2.44	2.44	2.43	2.31	2.50	2.33	2.31
Opposed electrodes							
Coated amount for cathode's outside (back surface) (g/m ²)	300	308	304	311	300	300	323
Coated amount for anode's inside (front surface) (g/m ²)	177	120	119	116	120	111	120
Ratio of coated amounts (cathode/anode)	2.56	2.57	2.55	2.68	2.50	2.70	2.69

The batteries produced in Embodiments and Comparative Embodiments were tested for an overcharge (3C-15V) and a discharge capacity. And, they were also tested for a charging and discharging cycle to measure the number of charging and discharging times (cycle life) until the capacity becomes 80% of the initial value. The results of the overcharge test were indicated by ⊙ for excellent, o for good, Δ for fair and x for bad. And, the discharge capacity was indicated with the battery capacity obtained in Comparative Embodiment 24 as reference. These test results are shown in Table 8.

Table 8

	Embodiments				Comparative Embodiments		
	48	49	50	51	25	26	27
Overcharge test	⊙	o	⊙	o	Δ	x	x
Comparison of capacity	+5	+3	+6	+2	±0	-3	-4
Cycle life (Frequency)	1400	1250	1500	1050	1000	800	750

It is apparent from Table 8 that in Embodiments 48 to 51 wherein the sheet electrodes having the coated amount of the electrode mix per unit area varied in a range of 2 to 10% between the front and rear surfaces were used for one or both of the cathode and the anode, the ratio of the coated amount of electrode mix between the cathode and the anode mutually opposed with the separator therebetween is well balanced at each section (the core and the outer periphery), safety against overcharge is improved and the discharge capacity is increased as compared with Compar-

ative Embodiment 24 wherein the electrode plate having the same coated amount on both surfaces is used. And, the charging and discharging cycling property was improved, and the cycle life was increased substantially.

On the other hand, in Comparative Embodiments 25 and 26 wherein the electrode plates having a difference in coated amount exceeding 10% between the front and back surfaces were used as the cathode or the anode, a ratio of coated amounts between the cathode and the anode mutually opposed with the separator therebetween is substantially different at each section, permeability of the electrolytic solution by charging or discharging is partly inferior, safety and capacity against overcharge are lowered, and cycle life is shortened.

It is apparent from the above description that the sheet electrode plate with the coated amount of electrode mix per unit area, namely the electrode active material density, varied between the front and back surfaces can be obtained by the present invention; the electrodes which are produced from the sheet electrode plate are wound into a roll with the separator therebetween and used as the cathode and/or anode, so that there can be obtained a nonaqueous electrolyte battery having excellent safety, high capacity, and an improved charging and discharging cycle property.

15 Claims

1. A method for manufacturing an electrode plate of a nonaqueous electrolyte battery comprising the steps of:
 running a sheet conductive base material in a first direction; and
 injecting an electrode material composition from a die nozzle onto a first surface of the running sheet conductive base material to form uncoated areas at predetermined intervals on the first surface along the first direction.
2. The method for manufacturing as set forth in claim 1, wherein the electrode material composition has an apparent viscosity of 500-100000 mPa·S (a shear rate of 13 sec⁻¹).
3. The method for manufacturing as set forth in claim 1, wherein the die nozzle is moved in a direction substantially perpendicular to and to separate from the first surface of the sheet conductive base material to form the uncoated areas.
4. The method for manufacturing as set forth in claim 1, wherein the sheet conductive base material is moved in a direction substantially perpendicular to and to separate from the tip of the die nozzle to form the uncoated areas.
5. The method for manufacturing as set forth in claim 1, wherein the die nozzle is moved along the first direction to form the uncoated areas.
6. The method for manufacturing as set forth in claim 1, wherein the electrode material composition is intermittently supplied from a coating liquid supply system to the die nozzle to form the uncoated areas.
7. The method for manufacturing as set forth in claim 1, wherein after the electrode material composition is coated onto the sheet conductive base material, the coated layer is dried and passed through a roller press, which has a pair of mutually opposed rollers disposed with a space therebetween, so as to be pressurized and compressed.
8. The method for manufacturing as set forth in claim 7, wherein a space D between the pair of mutually opposed rollers is in the following range:

$$d \times 0.6 \leq D \leq D_0$$

where, d is a thickness of the conductive base material, and D₀ is a target thickness of the compressed sheet electrode plate after pressurizing.

9. The method for manufacturing as set forth in claim 8, wherein the thickness d of the sheet conductive base material is 2 to 30 μm.
10. The method for manufacturing as set forth in claim 1, wherein the electrode material composition is injected from the die nozzle onto a first and/or second surface of the running sheet conductive base material to continuously decrease or increase at a predetermined ratio the coated amount per unit area along the first running direction of the sheet conductive base material within an area of the sheet conductive base material for a single battery.

11. The method for manufacturing as set forth in claim 10, wherein a difference of the coated amount per unit area at both ends on the coated section in the first direction of the sheet conductive base material is 2 to 20% with respect to the coated amount on the side with the decreased coated amount.
- 5 12. A method for manufacturing an electrode plate of a nonaqueous electrolyte battery comprising the steps of:
 running a sheet conductive base material in a first direction; and
 injecting an electrode material composition from a die nozzle onto first and second surfaces of the running
 sheet conductive base material to apply the electrode material composition per unit area in a different amount
 10 between the first area and the second area within an area of the sheet conductive base material for a single battery.
13. The method for manufacturing as set forth in claim 12, wherein a difference of the coated amount between the first surface and the second surface of the sheet conductive base material is 2 to 10% with respect to the coated
 15 amount on the side with the decreased coated amount.
14. An electrode plate of a nonaqueous electrolyte battery comprising:
 a rectangular sheet conductive base material having a first side, a second side which is shorter than the first
 20 side, a first surface, and a second surface; and
 an electrode material composition layer consisting of a first electrode material composition layer which is
 formed on the first surface of the sheet conductive base material and a second electrode material composition
 layer which is formed on the second surface with an uncoated area extending in a direction of the second side
 25 formed intermittently with respect to a direction of the first side at a position where the first electrode material
 composition layer and the second electrode material composition layer are mutually opposed.
15. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 14, wherein the electrode material
 composition uncoated area is formed on both ends of the electrode plate.
- 30 16. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 14, wherein the electrode material
 composition layer is formed by coating the electrode material composition onto the surface of the sheet conductive
 base material.
- 35 17. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 14, wherein the electrode material
 composition layer which is formed on at least one of the first and second surfaces of the sheet conductive base
 material has its coated amount continuously increases or decreased from one end to the other end along the first
 side of the sheet conductive base material.
- 40 18. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 17, wherein the coated amount of the
 electrode material composition increases or decreases by 2 to 20% from one end to the other end along the first
 side of the sheet conductive base material.
- 45 19. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 17, wherein the electrode material
 composition layer which is formed on at least one of the first and second surfaces of the sheet conductive base
 material has an active material density in the electrode material composition layer continuously increased or de-
 creased from one end to the other end along the first side of the sheet conductive base material.
- 50 20. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 14, wherein the first electrode material
 composition layer and the second electrode material coated layer are different in the coated amount of the electrode
 material composition.
- 55 21. The electrode plate of a nonaqueous electrolyte battery as set forth in claim 20, wherein a difference of coated
 amount per unit area of the electrode material composition between the first electrode material composition layer
 and the second electrode material coated layer is 2 to 10% with respect to the coated amount on the surface with
 the decreased coated amount.
22. An apparatus for manufacturing an electrode plate of a nonaqueous electrolyte battery to apply an electrode ma-
 terial composition onto a first surface and/or a second surface of a rectangular sheet conductive base material

which has a first side, a second side shorter than the first side, the first surface and the second surface, comprising:

means for moving the sheet conductive base material in a direction of the first side;

an injecting means which is disposed in the neighborhood of the surface of the moving sheet conductive base material and injects the electrode material composition in a direction substantially perpendicular to the first side;

a coating agent supplying means for intermittently supplying the electrode material composition to the injecting means;

a drying means for drying the electrode material composition coated onto the surface of the sheet conductive base material; and

a pressurizing and compressing means for pressurizing and compressing the dried sheet conductive base material and the electrode material composition.

23. The apparatus for manufacturing as set forth in claim 22, wherein the pressurizing and compressing means is a roller press which has a space between a pair of mutually opposed rollers.

24. The apparatus for manufacturing as set forth in claim 22, wherein a space D between the pair of mutually opposed rollers is in the following range:

$$d \times 0.6 \leq D \leq D_0$$

where, d is a thickness of the conductive base material, and D_0 is a target thickness of the compressed sheet electrode plate after pressurizing.

25. A nonaqueous electrolyte battery comprising:

a spiral cathode plate which is formed by winding a rectangular sheet conductive base material having a first side, a second side shorter than the first side, a first surface and a second surface around a shaft parallel to the second side;

a spiral anode plate which is formed by winding a rectangular sheet conductive base material having a first side, a second side shorter than the first side, a first surface and a second surface around a shaft parallel to the second side and which is disposed to be substantially parallel to the surface of the cathode plate;

an electrode material composition layer which is formed on the first and second surfaces of the cathode and anode plates and has an uncoated area, which is formed in multiple numbers on at least one of the first and second surfaces along the first side to extend in a direction of the second side, disposed intermittently with respect to the first side;

a separator which is held between the cathode plate and the anode plate; and

an outer member which has a positive terminal electrically connected to the uncoated area of the cathode plate and a negative terminal electrically connected to the anode plate.

26. The nonaqueous electrolyte battery as set forth in claim 25, wherein the electrode material composition uncoated area is formed on both ends of the electrode plate.

27. The nonaqueous electrolyte battery as set forth in claim 25, wherein the electrode material composition layer is formed by coating the electrode material composition onto the surface of the sheet conductive base material.

28. A nonaqueous electrolyte battery comprising:

a spiral cathode plate which is formed by winding a rectangular sheet conductive base material having a first side, a second side shorter than the first side, a first surface and a second surface around a shaft parallel to the second side;

a spiral anode plate which is formed by winding a rectangular sheet conductive base material having a first side, a second side shorter than the first side, a first surface and a second surface around a shaft parallel to the second side and which is disposed to be substantially parallel to the surface of the cathode plate;

an electrode material composition layer which is formed on at least one of the first and second surfaces of the cathode plate and/or the anode plate to have the coated amount continuously increased or decreased from one end to the other end along the first side of the cathode plate and/or the anode plate;

a separator which is held between the cathode plate and the anode plate; and

an outer member which has a positive terminal electrically connected to the cathode plate and a negative terminal electrically connected to the anode plate.

- 5 29. The nonaqueous electrolyte battery as set forth in claim 28, wherein the distance between the cathode plate and the anode plate is continuously decreased or increased from the center to the outer periphery of the battery.
30. The nonaqueous electrolyte battery as set forth in claim 29, wherein the distance between the cathode plate and the anode plate is increased or decreased by 2 to 20% from the center to the outer periphery of the battery.
- 10 31. The nonaqueous electrolyte battery as set forth in claim 28, wherein the electrode material composition layer which is formed on at least one of the first and second surfaces of the cathode plate and/or the anode plate has its active material density continuously increased or decreased from one end to the other end along the first side of the cathode plate and/or the anode plate.
- 15 32. The nonaqueous electrolyte battery as set forth in claim 31, wherein the active material density of the electrode material composition layer to be coated onto the cathode plate and/or the anode plate is increased or decreased by 2 to 20% from the center to the outer periphery of the battery.
- 20 33. The nonaqueous electrolyte battery as set forth in claim 28, wherein the electrode material composition layer is formed on the first surface and second surface of the cathode plate and/or the anode plate, and the coated amount of the electrode material composition is different between the first surface and the second surface.
- 25 34. The nonaqueous electrolyte battery as set forth in claim 33, wherein a difference of coated amount per unit area of the electrode material composition between the first electrode material composition layer and the second electrode material coated layer is 2 to 10% with respect to the coated amount on the surface with the decreased coated amount.

30

35

40

45

50

55

FIG. 1

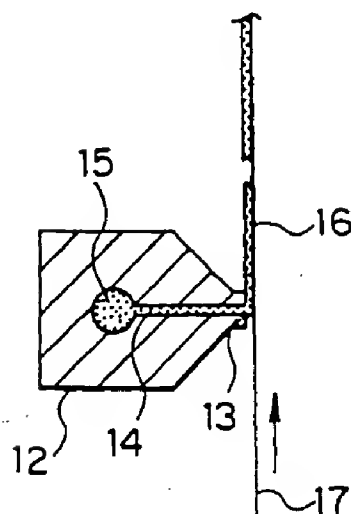


FIG. 2

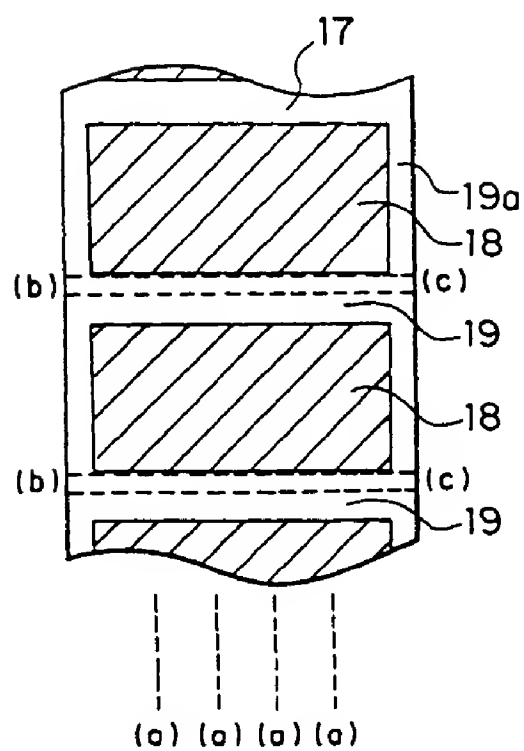


FIG. 3

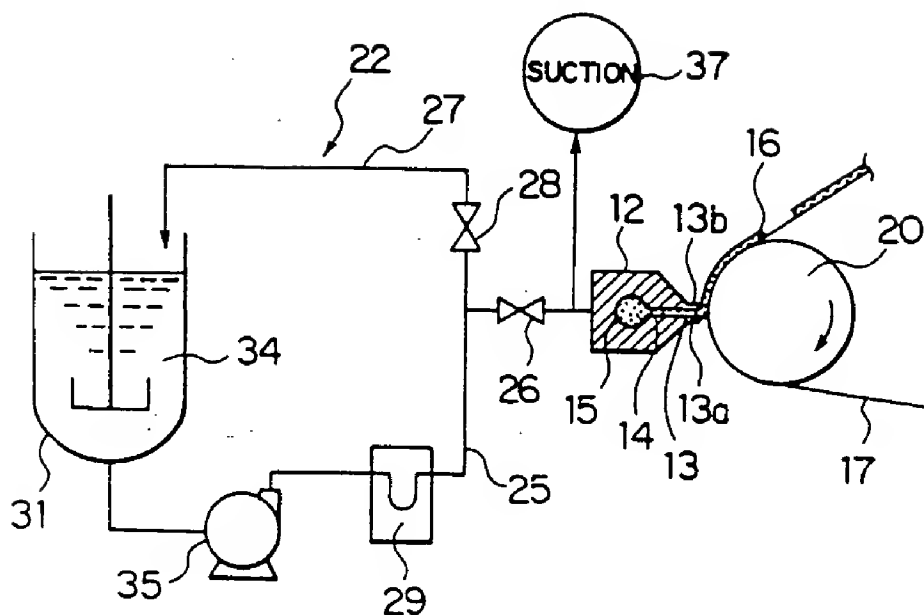


FIG. 4

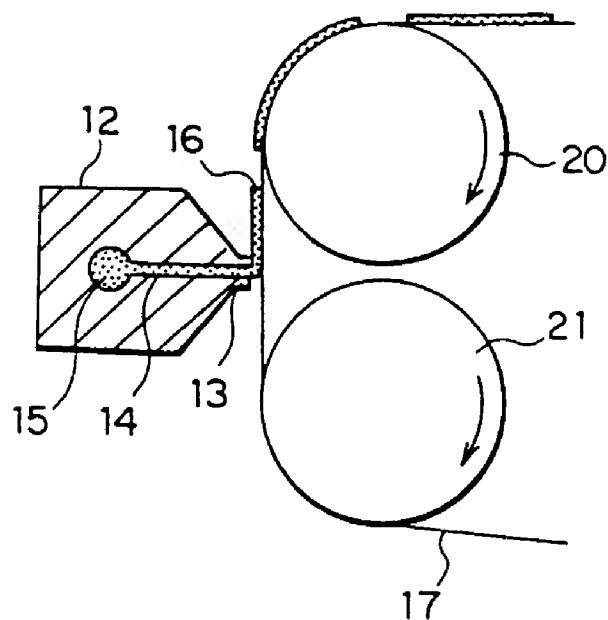


FIG. 5

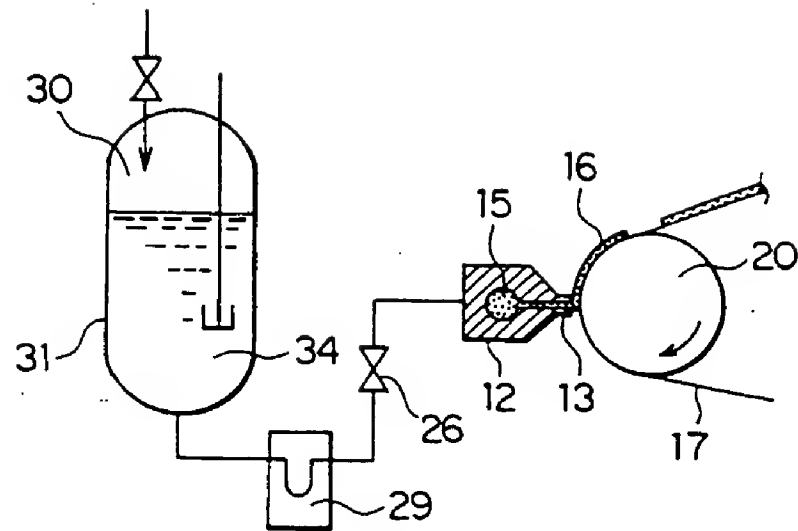


FIG. 6

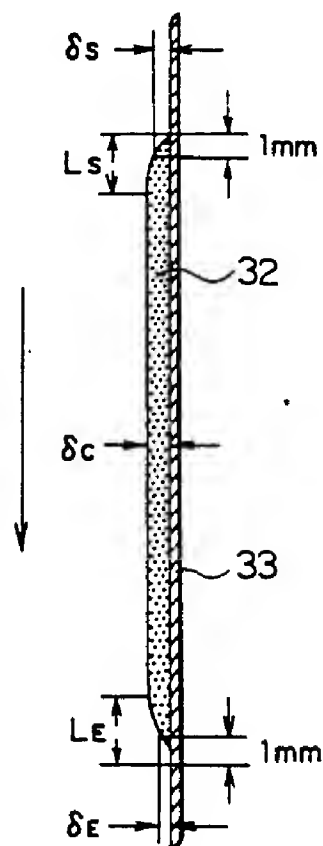


FIG. 7

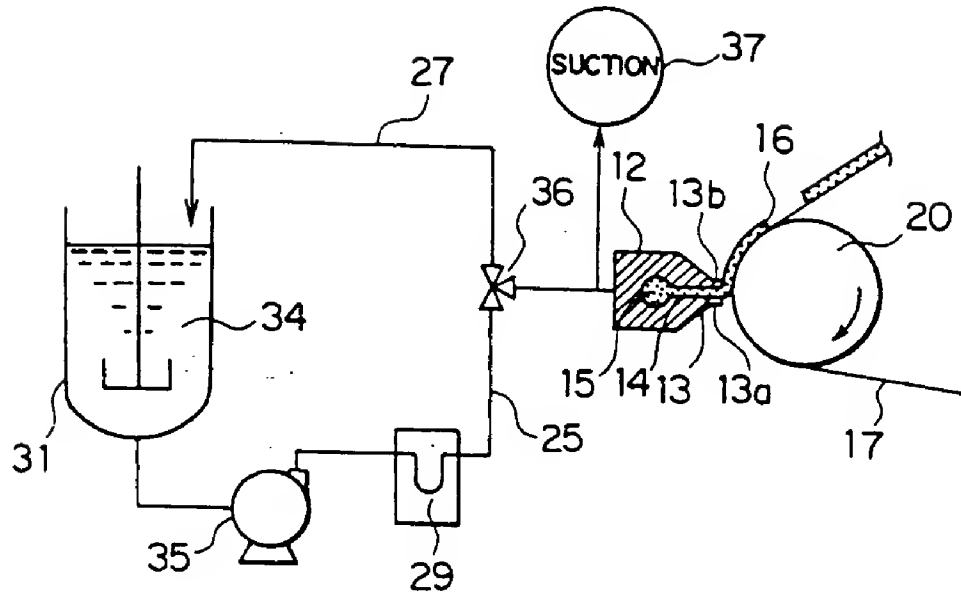


FIG. 8

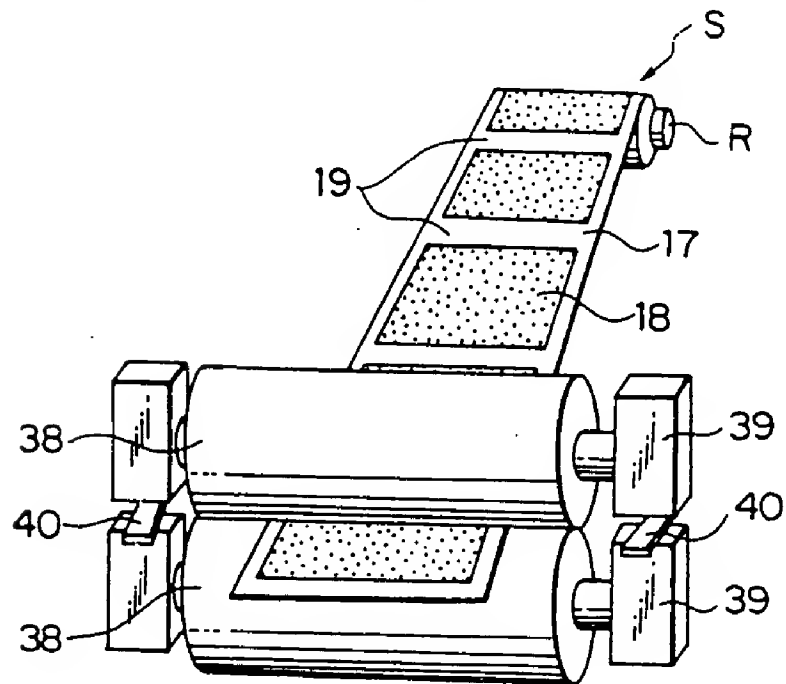


FIG. 9

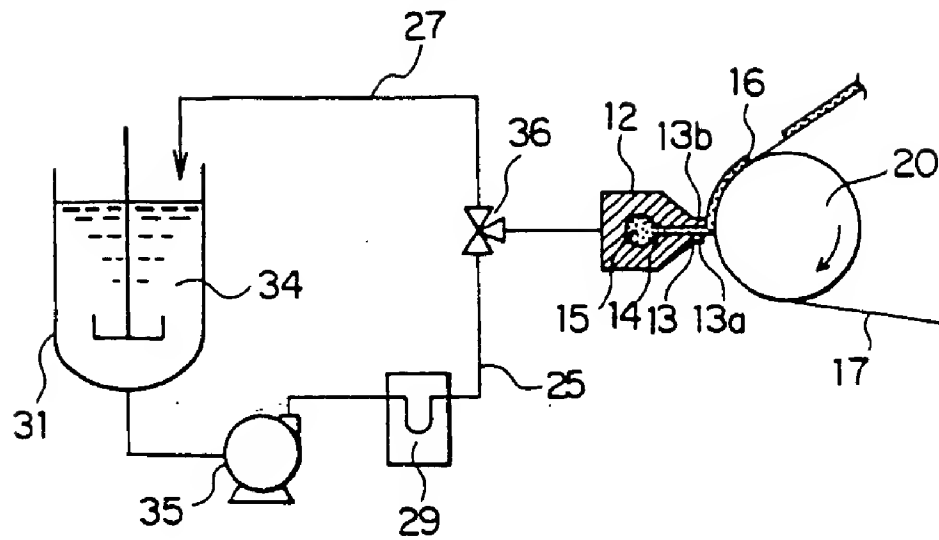


FIG. 10A

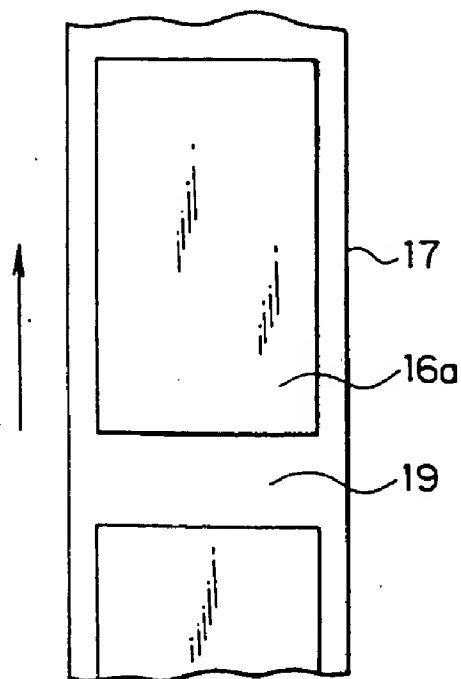


FIG. 10B

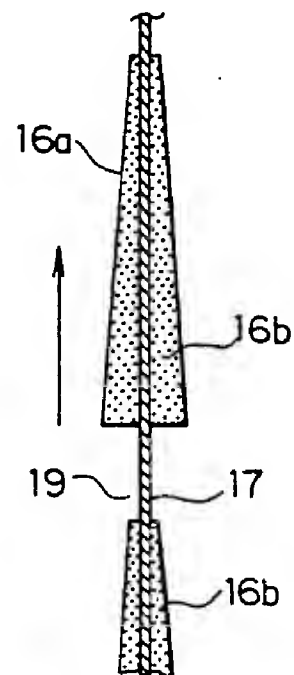


FIG. 11

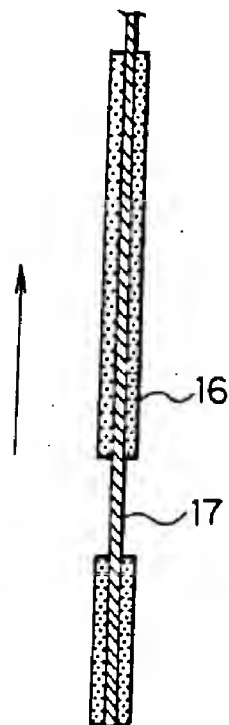


FIG. 12

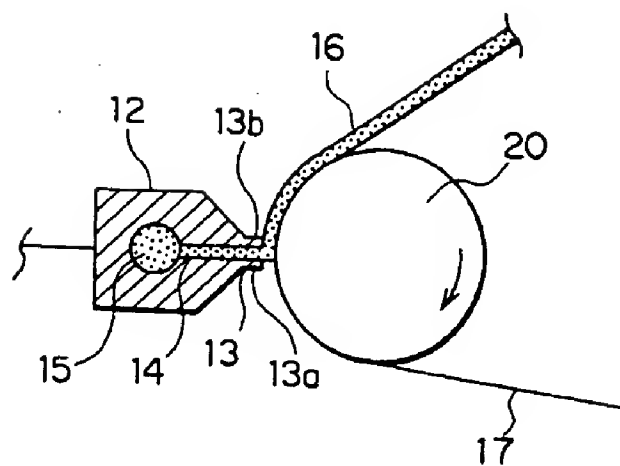


FIG. 13

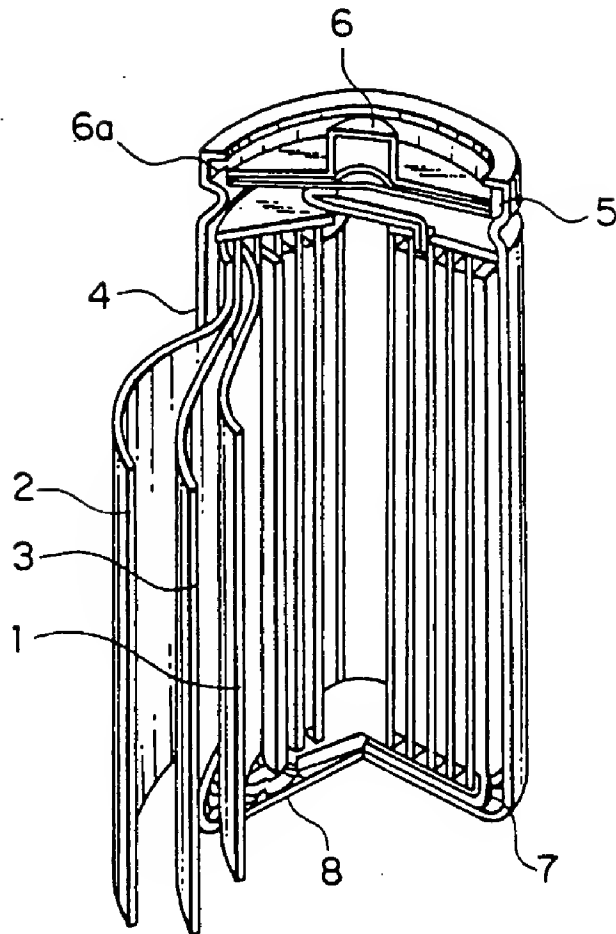


FIG. 14A

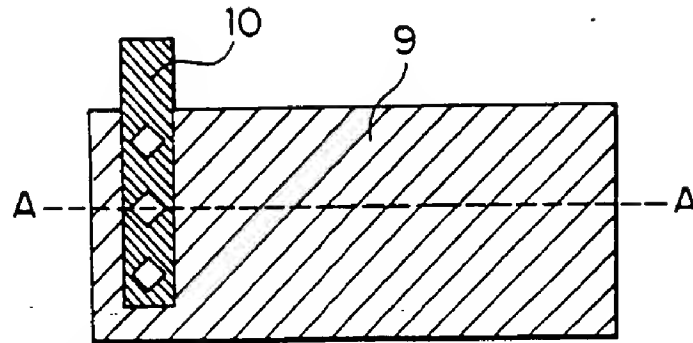


FIG. 14B

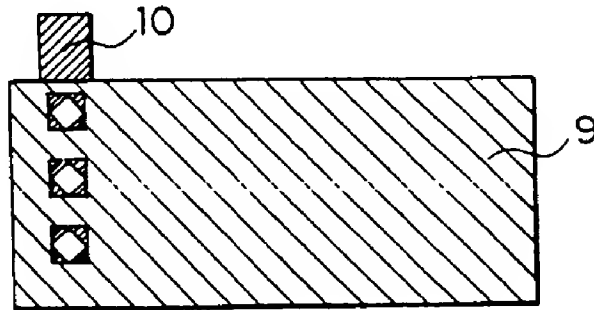


FIG. 14C

